TRANSITION METAL DIENYLS IN ORGANIC SYNTHESIS: THE PREPARATION OF 1.2-DIMETALLATED 1,3-DIENES **VIA "DOUBLE TRANSMETALLATION"**

Michael D. Fryzuk*, Charles Stone and Randy F. Alex Department of Chemistry, University of British Columbia **Vancouver, B.C., CANADA V6T lY6** and Raj. K. Chadha* Department **of Chemistry, University of Manitoba Winnipeg, Man., CANADA R3T 2N2**

Summary: A procedure is reported which provides for the preparation of functionalized dienes via transmetaUation OT protonation of zixconocene enyne complexes. A second transmetallation is possible to generate 1,2-disubstituted-1.3~dienes.

We have shown previously $1, 2$ that substituted enyne molecules can be converted to functionalized dienes having metal or metalloid substituents at the 1-position. The procedure involves a regioselective and chemoselective hydrozirconation³ step followed by a completely stereoselective transmetallation sequence to generate boron⁴, phosphorus⁴, selenium² and tin¹ dienes substituted at the 1-position. Because of the potential usefulness⁵ of heteroatom substituents both for activating the dienes toward Diels-Alder reactions and for subsequent elaboration, we were also interested⁶ in the preparation of the corresponding dienes with

To effect the goal of converting the readily available, enyne type starting materials to 2 substituted dienes, we developed an efficient strategy that utilizes the dienyl zirconium complexes 1-Zr as the starting materials. Utilization of the Buchwald procedure⁷ followed by either "double transmetallation" or protonation, produces the 2-substituted dienes that we sought with additional flexibility to introduce functionality at the l-position.

The dienyl zirconium reagents 1-Zr-a, b react with MeLi at -78^oC to generate⁷ the corresponding methyl-dienyl zirconocene derivatives $(\eta^5$ -C₅H₅)₂Zr(CH₃)CH=CH-CR=C@')R" **2a** and **2b;8 as these materials are** somewhat thennally labile, they were characterized in solution only and typically carried directly to the next step. Addition of PMe3 (8 - 10 equiv) results in a slow elimination of methane and the formation of the zirconocene alkyne complexes9 **3a** and 3b in good yields following the reported procedure.7

Since these formally Zr(II) alkyne derivatives can also be thought⁷ of as Zr(IV) metallocyclopmpene complexes, them existed the possibility that transmetallation might be accessible. In fact, this has proven to be quite facile, but the stereochemistry of this "transmetallation" was completely unanticipated.

Addition of trimethylstannyl chloride to either **3a** or **3b results in the** formation of a single isomeric product¹⁰ having, in each case, the tin at the 1-position and the zirconium at the 2-position; the relative stemchemistry of the Sn and Zr moieties was ambiguous by NOE difference spectroscopy, but determined to be *trans* by a single-crystal X-ray structure (singlecrystals of **E-1-Sn-2-Zr-b** were obtained from toluene/hexanes at -30^oC). Thus the sterochemistry of this first "transmetallation" step proceeds with *inversion*. The parent 2substituted zirconium dienyls can be accessed by reaction of **3a-b** with gaseous HCl(1 equiv) at -78oC, or more conveniently with pyridinium hydrochloride in tohtene.

preliminary fmdings hint at the potential of this strategy. With the zirconium in the two position, a subsequent "transmetallation" can be effected in analogy to previous studies. $1,2,4,11$ Addition of Me₂PCl to E-1-Sn-2-Zr-a,b cleanly produces the stannyl phosphines E-1-Sn-**2-P-a,b.** In addition, the first "transmetallation" also proceeds with Ph₂BBr or Me₂PCl to generate the corresponding **E-l-B-2-Zr-a** and **E-l-P-2-Zr-b dienes.**

These results collectively illustrate the flexibiity of this "double transmetallation" procedure for the preparation of 1,2-dimetallakd-1,3dienes with a wide variety of substituents. The mechanism involved in the reaction of $3a-b$ with both HX and $L₁MX$ systems, and the detailed reactivity of these new, highly functionalized dienes are currently under investigation.

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- 8. **2a:** ¹H NMR (C₆D₆, 400 MHz, C₆D₅H at 7.15 ppm) δ 0.05 (s, 3H, CH₃), δ 3.17 (s, 3H, OCH₃), δ 5.46 (dd, J_{DC}= 12 Hz, J_{BC}= 10 Hz, 1H, H_C), δ 5.80 (s, 10H, Cp), δ 6.37 (dd, J_{AB}= 19 Hz, J_{BC}= 10 Hz, 1H, H_B), δ 6.58 (d, J_{CD}= 12 Hz, 1H, H_D), δ 6.98 (d, J_{AB}= 19 Hz, 1H, H_A). 2b: ¹H NMR (C₆D₆, 400 MHz, C₆D₅H at 7.15 ppm) δ 0.08 (s, 3H, CH₃), δ 1.52 (m, 2H, H_{E/E}), δ 1.60 (m, 2H, H_E/E), δ 2.06 (m, 2H, H_{D/D}), δ 2.14 (m, 2H, H_D/_D), δ 5.72 (bm, 1H, H_C), δ 5.80 $(s, 10H, Cp), \delta 6.47$ (d, J_{BA}= 19 Hz, 1H, H_B), $\delta 6.95$ (d, J_{AB}= 19 Hz, 1H, H_A).
- 9. 3a: ¹H NMR (C₆D₆, 400 MHz, C₆D₅H at 7.15 ppm) δ 0.98 (d, J_{PH}= 5.9 Hz, 9H, PCH₃), δ 3.47 (s, 3H, OCH₃), δ 5.32 (d, J_{PH}= 1.6 Hz, 10H, Cp), δ 7.01 (dd, J_{BC}= 12 Hz, J_{AB}= 1.4 Hz, 1H, H_B), δ 7.30 (d, J_{BC}= 12.2 Hz, 1H, H_C), δ 7.39 (dd, J_{PH}= 4.1 Hz, J_{AB}= 1.4 Hz, 1H, H_A). ³¹P^{{1}H}</sub> NMR $(C_6D_6, 121.5 \text{ MHz})$ δ 0.8 (s). Anal. calc. for C₁₈H₂₅OPZr: C, 56.96; H, 6.64. Found: C, 57.15; H, 6.60. 3b: ¹H NMR (C₆D₆, 400 MHz, C₆D₅H at 7.15 ppm) δ 0.98 (d, J_{PH} = 5.8 Hz, 9H, PCH₃), δ 1.75 (m, 2H, H_{D/D}), δ 1.84 (m, 2H, H_{D/D}), δ 2.47 (m, 2H, H_{C/C}), δ 2.69 (m, 2H, H_{C/C}), δ 5.34 (d, Jp_H= 1.7 Hz, 10H, Cp). δ 6.27 (m, 1H, H_B), δ 7.53 (d, J_{PH}= 4.1 Hz, 1H, H_A). ³¹P[¹H] (C₆D₆, 121.5 MHz, 85% H₃PO₄ at 0 ppm) δ 0.4 (s). Mass spectrum (EI, 70eV) calc. for C₂₁H₂₉PZr (P⁺⁾, m/e 402. Found, m/e 326 (P-PMe3)⁺, m/e 220 (ZrCp₂)⁺, and m/e 546 (very weak) (P-PMe3+ZrCp₂) Anal. calc. for $C_{21}H_{29}PZr$: C, 62.49; H, 7.24. Found: C, 62.31; H, 7.38.
- 10. 1-Sn-2-Zr-a: ¹H NMR (C₆D₆, 400 MHz, C₆D₅H at 7.15 ppm) δ 0.27 (s with ¹¹⁷Sn and ¹¹⁹Sn satellites, $J(117Sn-H)= 52 Hz$, $J(119Sn-H)= 54.4 Hz$, $9H$, $SmMe₃$), δ 3.33 (s, 3H, OCH₃), δ 5.76 (s, 10H, Cp), δ 6.38 (dd, J_{DC}= 12.5 Hz, J_{AC}= 1.4 Hz, 1H, H_C), δ 6.64 (d, J_{CD}= 12.5 Hz, 1H, H_D), δ 7.56 (d with single set of Sn satellites, $J_{AB} = 1.4$ Hz, $J_{SnH} = 24.5$ Hz, 1H, H_A). Anal. calc. for C₁₈H₂₅ClOSnZr: C, 43.00; H, 5.01. Found: C, 43.31; H, 5.07. 1-Sn-2-Zr-b: ¹H NMR (C₆D₆, 400 MHz, C₆D₅H at 7.15 ppm) δ 0.23 (s with ¹¹⁷Sn and ¹¹⁹Sn satellites, $J(117Sn-H)= 52 Hz$, $J(119Sn-H)= 54.4 Hz$, $9H$, $SmMe₃$), δ 1.55 (m, 2H, $H_{E/E'}$), δ 1.63 (m, 2H, H_{E'/E}), δ 1.98 (m, 2H, H_{D/D}'), δ 2.11 (m, 2H, H_{D'/D}), δ 5.28 (m, 1H, H_C), δ 5.78 (s, 10H, Cp), δ 7.22 (s with single set of Sn satellites, J_{SnH} 43 Hz, 1H, H_A). Mass spectrum (EI, 70eV) calc. for C₂₁H₂₉ClSnZr (P⁺), m/e 526. Found, m/e 526 (weak), m/e 511 (P-PMe3)⁺, m/e 420 (P- $C_6H_{11}Cl$)⁺. Anal. calc. for $C_21H_{29}ClSnZr$: C, 47.88; H, 5.55. Found: C, 48.20; H, 5.74.
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